Sl.No: M21927 Course Code: BP104T

VINAYAKA MISSION'S RESEARCH FOUNDATION (DEEMED TO BE UNIVERSITY), SALEM

B.PHARM. DEGREE EXAMINATION – JANUARY 2020 First Semester

PHARMACEUTICAL INORGANIC CHEMISTRY

Time: Three hours

Maximum: 75 marks

I. Write essays on any **TWO** questions:

 $(2 \times 10 = 20)$

- 1. Write the principle, procedure and reaction involved in the limit test for a) Lead b) Iron
- 2. Describe preparation assay and uses of hydrogen peroxide.
- 3. What are buffers? Give a list of official buffers. Explain the role of buffer system in pharmaceutical preparation.
- II. Write short answers on any **SEVEN** questions:

 $(7 \times 5 = 35)$

- 4. Explain the role of fluoride in dental caries.
- 5. Write the method of preparation, properties and uses of sodium bicarbonate
- 6. What are astringents? Write molecular formula properties and uses of potash alum.
- 7. Define haematinics. Explain the properties and method of preparation of ferrous gluconate.
- 8. Write the storage and handling of radiopharmaceuticals.
- 9. Outline the role of activated charcoal in poising.
- 10. Write method of preparation and uses of copper sulphate and sodium thiosulphate.
- 11. Explain the various type of radiation.
- 12. ORS (oral rehydration salt).

III. Write short notes on:

 $(10 \times 2 = 20)$

- 13. What is meant by limit test?
- 14. Define acid and base.
- 15. What are antacid?
- 16. Give the molecular formula a) Sodium orthophosphate b) Sodium Nitrite
- 17. Radioisotopes.
- 18. Give the preparation and use of magnesium sulphate.
- 19. Write a note onpotassium Permanganate.
- 20. Define Emetics.
- 21. Ideal properties of antacid.
- 22. Pharmacoopoeia.
